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(54) Title: SALTS OF 2-(3-BENZOYLPHENYL)PROPIONIC ACID WITH ACHIRAL AND CHIRAL ORGANIC BASES AND PHARMACEUTICAL COMPOSITIONS THEREOF

#### (57) Abstract

The salts of S(+) 2-(3-benzoylphenyl)propionic acid and of R(-) 2-(3-benzoylphenyl)propionic acid with an achiral, organic base such as tris-(hydroxymethyl)aminomethane or a chiral organic base such as D-lysine, L-lysine, L-arginine, (R) 3-(4-phenylpiperazin-1-yl)propane-1,2-diol and (S) 3-(4-phenylpiperazin-1-yl)propane-1,2-diol, the process for their preparation and the corresponding pharmaceutical compositions containing said salts are described.

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#### Description

# Salts of 2-(3-benzoylphenyl)propionic acid with achiral and chiral organic bases and pharmaceutical compositions thereof

#### Technical Field

The object of the present invention relates to salts of 2-(3-benzoylphenyl)propionic acid with achiral and chiral organic bases, and to the pharmaceutical compositions containing them.

A further object of the invention relates to the process for the preparation of said salts.

More particularly, the present invention relates to the salts of the S(+) and R(-) enantiomers of 2-(3-benzoylphenyl)propionic acid with achiral amine, such as, for example, tris-(hydroxymethyl)aminomethane, also known as tromethamine, and with chiral amine such as, for example (R) and (S) 3-(4-phenylpiperazin-1-yl)propane-1,2-diol, also known as dextrodropropizine and levodropropizine, and with basic \( \times \)-aminoacids such as, for example, D-lysine, L-lysine and L-arginine, all salts which may be separated as single chemical individuals of high optical purity.

## Background of the Invention

- Because of its high tolerability, the (S,R) (±) 2-(3-benzoylphenyl)propionic acid, also known as ketoprofen, is one of the non-steroidal anti-inflammatories of widespread use in clinics, both for the treatment of serious inflammatory conditions and for its use as an anagelsic and antipyretic.
- 25 Pharmaceutical compositions of current use containing

ketoprofen, have racemate as its active principle, where the two enantiomers S(+) and R(-) are present in equimolecular ratio between themselves.

The active principle is normally used as free acid, practically insoluble in water, in pharmaceutical compositions destined for oral use, while for alternative ways of administration, for example that of parenteral administration, adaptable ketoprofen salts with organic and inorganic bases are used.

In the past, all the pharmacological activities peculiar to the racemate of 2-arylpropionic chiral acid, were thought to be constitutive of the enantiomer S(+) which only was found to inhibit the endogenous synthesis of the pro-inflammatory algogene and pirogene prostaglandines, in which respect the antipode R(-) is inactive or practically so. On the other hand, it is well known that the R(-) enantiomer of the 2-arylpropionic acids undergoes, to a variable extent and in a way animal species dependent, metabolic epimerization in the S(+) enantiomer, an event which, for a long time, has prevented a correct characterization of the pharmacological properties of the individual enantiomers.

Only recently, using flurbiprofen, a chiral 2-arylpropionic anti-inflammatory and analgesic acid, whose enantiomers are not metabolically converted one into another, K. Brune et al.

(Exprientia, 47, 257, 1991) have clearly shown that the inhibition of the prostaglandine synthesis mainly mediates the anti-inflammatory activity of the compound, while mechanisms independent from the inhibition of the prostaglandine synthesis contribute to the analgesic effects of the racemate.

30 Of the two antipodes, the S(+) form inhibits the

prostaglandine synthesis, the inflammation and the perception of the pain, while the R(-) antipode, which has much less effect on the inhibition of the prostaglandine synthesis and has no effect on the inflammation, blocks the perception of the pain with a potency rather similar to that of the antipode S(+).

S(+) flubiprofen is clearly ulcerogenic for the gastroenteric mucose, unlike the R(-) enantiomer. On the basis of these results, the A.A.s conclude on the existence of additional mechanisms of analgesia and propose a new and correct therapeutic use of the R(-) 2-arilpropionic acids as analgesics.

These concepts are further enphatized in a successive article (K. Brune et al., J. Clin, Pharmacol., 32, 944, 1992) where it is concluded that, having recourse to the use of individual enantiomers of the chiral 2-arilpropionic acids instead of the racemate, it is possible:

- a) to reduce the dose and by that the metabolic load;
- b) to reduce the variability in clinical response by eliminating the biochemical inversion pathway;
- c) to reduce compliance problems due to unnecessarily high doses;
- d) to establish more specific drug treatment (R-enantiomers in occasional pain, S-enantiomers in rheumatic disorders).

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#### Description of the invention

The object of the present invention relates to pharmacologically active salts of 2-(3-benzoylphenyl)-propionic acid with achiral and chiral organic bases and to the process of their preparation and to the pharmaceutical

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compositions containing them.

More particularly those are salts of the enantiomeric forms S(+) and R(-) of the 2-(3-benzoylphenyl)propionic acid with achiral amines such as. for example. tris(hydroxymethyl)aminomethane, also known as tromethamine, and with chiral amines such as, for example, (R) and (S) 3-(4-phenylpiperazin-1-yl)propane-1,2-diol, also known dextrodropropizine and levodropropizine, and with basic  $\propto$  aminoacids such as, for example, D-lysine, L-lysine and L-arginine, salts which may all be isolated as single chemical individuals having high optical purity.

The salts of S(+) 2-(3-benzoylphenyl)propionic acid with the above-mentioned bases are in particular usefully employed in the treatment of those pathological symptoms of rheumatoid and cronic type, which require the drug to be administered at high dosage, continuously and for long periods of time.

In such event, the presence in the racemic form of the enantiomer R(-), which is ineffective as an inflammatory drug, would represent for the patient an unnecessary metabolic load which would even be harmful. In fact the optical antipode R(-), which is pharmacologically inactive in inhibiting the prostaglandine synthesis, and therefore as anti-inflammatory agent, does not or only very slightly and in a kinetically and therapeutically inefficient way, undergo epimerization in man to the enantiomeric form S(+) to which the anti-inflammatory activity of the racemate are due.

The salts of the R(-) 2-(3-benzoylphenyl)propionic acid with the above-mentioned bases are in particular usefully employed in treating acute painful symptoms of spastic type (renal, biliary or hepatic colics) and/or tissue-type characterized by

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sensibilization of the nerve ends and/or of traumatic type.

More generally and in some situations of acute pain, the same compounds could be proposed as a true alternative to the use of narcotics.

- It is important and desirable that for the treatment of acute and very painful manisfestations, there are pharmaceutical compositions suitable for immediate use and manageable, which rapidly release the active principle and are of high bio-availability.
- Typical examples of these compositions are those by parenteral administration and/or by oral administration which are drinkable, which allow a fine dispersion of the active principle. Due to the scarce solubility in water of the active principle, it is necessary to resort, for these purposes, to the use of salts, as single chemical individuals or obtained by extemporary salification during the pharmaceutical formulation process.

Pharmaceutical formulations are known which contain salts of racemic ketoprofen and those containing sodium salt (ketoalgine  $^{R}$ ) and D,L-lysine salts (Artrosilene  $^{R}$ ) are of current use.

More recently, in patent applications WO 93/16689 (2802, 1992) and WO 93/17677 (09.03.1992) relating to the use of R(-) ketoprofen as an analgesic, pharmaceutical compositions containing as active principle R(-) ketoprofen or a salt thereof with pharmaceutically acceptable organic and inorganic non-toxic bases, are indicated. In both cases, a general reference is made to addition salts of R(-) ketoprofen with various metal ions among which those with alkaline and earth-alkaline metals and with various organic bases, among

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which the salts with the basic amino acids, such as lysine and arginine.

While the salification process of a chiral 2-arylpropionic acid, in the racemic form, does not involve problems concerning the chemical racemizations of the active principle, this aspect assumes a noticeable relevance when the salification involves the same chemical species but in their optically active form.

In the latter case, the possibility of an oncoming chemical racemization during the salification, drying and storage processes of the raw material, or successively in a state of solution, or during manipulation of the pharmaceutical formulation, cannot be excluded.

It follows that the salification process, the characteristic
of the chemical specie salt, the more appropriate to preserve
the integrity of the active principle, are not accessory
elements of the manipulation and of the practical utilization
of the enantiomerically pure active principle.

The salts of the R(-) 2-(3-benzoylphenyl)propionic acid or 20 R(-) ketoprofen, the salts of the S(+) 2-(3-benzoylphenyl)propionic acid or S(+) ketoprofen with achiral organic bases, such as for example, tromethamine, or with chiral, enantiomerically pure, organic bases, such as L-lysine, D-lysine, L-arginine, (S)

3-(4-phenylpiperazin-1-yl)propane-1,2-diol and (R)
3-(4-phenylpiperazin-1-yl)propane-1,2-diol have been obtained as single chemical individuals, and form the object of present invention.

The process for preparing the above said salts consists of a salification reaction, in a suitable solvent and kept warm, of

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powder by X-ray.

one of the above-mentioned bases, with R(-) 2-(3-benzouylphenyl)propionic acid or S(+) 2-(3-benzoylphenyl)propionic acid, having an enantiomeric purity of no less than 95%. After cooling, the corresponding salts separate themselves in a good yield, as such or after re-crystallization, and contain the salifying acid which has an optical purity of no less than 99%.

Preferred solvents used in the salification reaction are alcohols such as methanol, ethanol, propanol and isopropanol;

10 ketones such as acetone; water and /or mixtures containing such solvents.

In the salification process with one of the above mentioned  $\alpha$ -aminoacids, specifically in the case of lysine, the solvent more particularly preferred is aqueous isopropanol, in a ratio

acid:solvent of 1 to 20, with an average water content of 3%. In these experimental conditions the salification, for example, of the R(-) 2-(3-benzoylphenyl)propionic acid with L-lysine gives crystalline solids which are easily filtered and which, after drying, allow to isolate single crystalline individuals of high purity and stability, which may be characterized by I.R. spectrometry and by diffraction of the

The salts of the enantiomers of the S(+) and R(-) 2-(3-benzoylphenyl)propionic acids of the present invention, are stable solids, easily filtered and obtainable during the phase of production or purification. They can be in the form of amorphous solids only apparently crystalline, such as the salts of S(+) ketoprofen with L-arginine and of R(-) ketoprofen with D-lysine, or in the form of a crystalline monohydrate such as the salt of S(+) ketoprofen with D-lysine.

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The salt of R(-) ketoprofen with L-lysine is one with a residual humidity of about 1%, which in time does not absorb hydration water, and it keeps itself stable in time and is, therefore, particularly manageable, either as such, or as a pharmaceutical composition in which it is contained.

The enantiomeric forms R(-) and S(+) of the 2-(3-benzoylphenyl) propionic acid, or R(-) and S(+) ketoprofen, of convenient optical purity are obtained by optical resolution of the  $S,R(\pm)$  ketoprofen.

- 10 In particular, R(-) ketoprofen is preferably obtained through a process which utilizes the salification of (R,S) ketoprofen, at room temperature, with (S) 3-(4-phenylpiperazin-1yl)propan-1,2-diol in acetone at relatively high dilutions (acid:solvent=1:15). After filtration of a salt, the 15 enantiomerically rich in S(+) ketoprofen and cooling the waters to 0°C, the R(-)ketoprofen 3-(4-phenylpiperazin-1-yl)propane-1,2-diol salt crystallizes, a highly satisfactory optical purity. As alternative, the salification at 40°C, methanol in (acid:solvent = 1 g:5 ml) with R(+) 3-(4-phenylpiperazin-1-20 yl)propane-1,2-diol produces crystallization on cooling of the R(-) ketoprofen with R(+)3-(4-phenylpiperain-1yl)propane-1,2-diol having an optical purity of about 80%. The desired 98% optical purity is reached by recrystallization
- 25 from acetone or by successive treatment with S(-) 3-(4-phenylpiperazin-1-yl)propane-1,2-diol in acetone (solute:solvent = 1:10).
  - S(+) 2-(3-benzoylphenyl)propionic acid having enantiomeric purity (o.p.) of no less than 90% is obtained, at first time, by salifying the racemate  $S_*R(\pm)$  2-(3-benzoylphenyl)propionic

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acid in acetone with R(-) 3-(4-phenylpiperazin-1-yl)propane)-1,2-diol. A diastereoisomer salt crystallizes which, after filtration and drying in vacuo, is suspended in water. After acidification of the suspension and extraction with an organic solvent such as, for example, ethyl ether, cyclohexane and/or mixtures thereof, the S(+) 3-(2-benzoylphenyl)propionic acid is obtained with a yield of 60  $\pm$  5%, having an optical purity of at least 90%.

A further improvement in enantiomer yield is coming by a resolution process that uses salification of the racemic acid with half molecular equivalent of the resolvent (S) or (R)-dropropizine.

In comparison to the known salts in which 2-(3-benzoylphenyl)propionic acid is contained in racemic form, the salts of the present invention show a higher purity degree and a greater stability which positively reflects on the handling of the product as such or as a pharmaceutical preparation containing it. In particular, in the case where the salts are formed with the D- and L-lysine enantiomers, the presence of a certain quantity of crystallization water or humidity allows a higher stability of the products.

Moreover the salts of the invention offer the advantage of allowing the preparation of pharmaceutical compositions, the active principle of which is constituted by diastereoisomerically pure single molecular individualities that, as such, give an absolute consistency of quality even with the changing of the preparation batch.

The salts of the invention may be suitably mixed with pharmaceutically acceptable excipients and formulated in a suitable manner for oral, intranasal, parenteral, topical and

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inhalant administration. The pharmaceutical compositions, which contain as active principle an effective quantity of one or more salts of the enantiomer S(+) or R(-) 2-(3-benzoylphenyl) propionic acid with an organic achiral base such as tromethamine and/or an organic chiral base selected among L-lysine, D-lysine, L-arginine (S) and (R) 3-(4-phenylpiperazin-1-yl)propane-2,3-diol may be in the form of pills, tablets, dragées, granulates, powders, emulsions, solutions, foams, creams, suppositories and spray.

The quantity of the active principle evaluated as salifying acid which is daily administered may vary depending on the type of the administration chosen, on the age and on the condition of the patient.

In the case of oral administration it varies from 20 to 200 mg which may be divided in several doses or as a long-lasting single dose and, in the case of injectable administration, it varies from 10 to 100 mg which may be divided in several doses. For topical administration concentrations of 1% to 10% are suitable, while in the case of sublingual administration single doses of 10 to 50 mg up to a daily total dose of 200 mg may be administered. For the aerosol administration single doses of 10 to 100 µg up to a daily total dose of a maximum of 800 µg may be administered.

Pharmaceutical formulations suitable for the administration of the salts of the invention as nasal spray in concentration of from 0,1 to 2% and those suitable as colluttory in concentration of from 5 to 15%.

1. Preparation of R(-) 2-(3-benzoylphenyl)propionic acid

To a solution of 400 g (R,S)-2-(3-benzoylphenyl)propionic acid

in 8 l acetone are added, under stirring and maintaining the

temperature at 20-25°C by means of external cooling, 440 g S(-) 3-(4-phenylpiperazin-1-yl)propane-1,2-diol. Stirring is maintained for a further 15 minutes until complete dissolution then the salt is allowed to crystallize. After 6 hours the precipitate is filtered, dried in the air and 370 g (25,2'S') 3'-(4'-phenylpiperazin-1-yl)propane-1',2'-diol 2-(3-benzoyl-phenyl)propionate are obtained.

 $[\propto]_{D} = -2.8^{\circ} \text{ (MeOH, o.p.(S) 82\%)}$ 

The mother waters are concentrated to a volume of 6 l and 10 cooled to 0°C and separate 280 g (2R,2'S) 3'-(4'-phenylpiperazin-l'-yl)propane-l',2'-diol 2-(3-benzoylphenyl)-propionate.

 $[\propto]_{D} = -19.8^{\circ} \text{ (MeOH, o.p. (R) 97.98\%)}$ 

Recrystallization from acetone of the compound (solute:solvent 1:10) gives the enantiomerically pure salt, melting at 107-109°C.

 $\left[\propto\right]_{D} = -20.8^{\circ} \text{ (MeOH)}$ 

A suspension of 25 g (2R,2'S) 3'-(4'-phenylpiperazin-1'-yl)propane-1',2'-diol 2-(3-benzoylphenyl)
20 propionate in 30 ml water is acidified to pH 1 with 2N sulphuric acid, then twice extracted with 4 ml ethylacetate. The organic phases are collected together, washed with water, made anhydrous on sodium sulphate and evaporated to dryness. By recrystallization of the residue from cyclohexane 11 g R(-)

25 2-(3-benzoylphenyl)propionic acid, melting at 75-76°C are obtained.

 $[\propto]_D = -51^\circ (1\% \text{ in } CH_2 Cl_2)$ 

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2. Preparation of S(+) 2-(3-benzoylphenyl)propionic acid

Grams 22 of (R,S) 2-(3-benzoylphenyl)propionic acid are treated with 20 g of R(+) 3-(4-phenylpiperazin-1-yl)propane-

1,2-diol in 0.1 1 methanol and 18 g of (2R,2'R) 3'-(4'-phenylpiperazin-1'-y1)propane-1',2'-diol 2-(3-benzoylphenyl)-propionate are obtained.

 $[\infty]_{D} = +2.9^{\circ} \text{ (MeOH, o.p. (R) 80%)}.$ 

Removing by distillation the solvent and crystallizing the residue from 250 ml acetone, 10 g of (2S,2'R) 3'-(4'-phenyl-piperazin-l'-yl)propane-l',2'-diol 2-(3-benzoylphenyl)-propionate, are obtained.

 $[\propto]_{D} = +20^{\circ}$  (MeOH, o.p. (S) 98%).

The product is dissolved in water and acidified to give S(+) 2-(3-benzoylphenyl)propionic acid melting at 74-77°C.

 $[\infty]_D = +51.2^{\circ} (1\% \text{ CH}_2 \text{ Cl}_2)$ 

Hereunder are some Examples for a better illustration of the invention.  $\dot{}$ 

15 Example 1

R(-) 2-(3-benzoylphenyl)propionic acid L-lysine salt R(-) 2-(3-benzoylphenyl)propionic acid D-lysine salt

Grams 300 of R(-) 2-(3-benzoylphenyl)propionic acid are dissolved at room temperature in 3 1 of isopropanol.

- The solution is heated, under stirring, to 60°C and a solution of 168 g L-lysine in 160 ml of deionized water are added thereto. The solution is filtered hot, diluted, under stirring, with 3 l of isopropanol and left to cool. When the crystallization begins at 48-50°C the stirring is interrupted.
- Two hours later a crystalline precipitate is filtered, washed with 600 ml isopropanol. It is dried in the air; after sieving on a 500  $\mu$  sieve it is dried in vacuo at 50°C (20 mm Hg). Grams 390 of R(-) 2-(3-benzoylphenyl)propionic acid L-lysine salt, melting at 106-108°C are obtained. The X-ray diffraction spectrum is given in Figure 1, are obtained.

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(H<sub>2</sub>0)K.F.: 1.4%

 $[x]_D = +10.6^{\circ} \text{ (c=1\%, MeOH); } [x]_{436} = +30.4^{\circ} \text{ (c=1\% MeOH)}$ Operating in a similar manner, salifying with D-lysine R(-) 2-(3-benzoylphenyl)propionic acid D-lysine salt melting at  $106-108^{\circ}\text{C}$ , as amorphous solid was obtained.

 $[\propto]_D = +1.2^{\circ} (c=1\%, MeOH); [\sim]_{436} = +10.4^{\circ} (c=1\% MeOH)$ Example 2

R(-) 2-(3-benzoylphenyl)propionic acid tris-hydroxymethyl-methylammonium salt

10 S(+) 2-(3-benzoylphenyl)propionic acid tris-hydroxymethyl-methylammonium salt

A solution of 5 g R(-) 2-(3-benzoylphenyl) propionic acid in isopropanol is treated with a solution of 2.4 g of tris-hydroxymethylaminomethane in 2.5 ml deionized water. It is evaporated with great care under vacuo and the oily residue

taken up with 20 ml of ethylether. The crystalline solid which is separated is filtered and it gives 5.4 g of R(-) 2-(3-benzoylphenyl)propionic acid tris-hydroxymethylmethyl-ammonium salt, melting at 101-103°C.

20 (H<sub>2</sub>0)K.F.: 2.05%

 $[\alpha]_D$  = +4° (c=1%, MeOH);  $[\alpha]_{436}$  = +18.2° (c=1% MeOH) Operating in a similar manner, by salifying the S(+) 2-(3-benzoylphenyl)propionic acid the S(+) 2-(3-benzoylphenyl)-propionic acid tris-hydroxymethylmethylammonium salt, melting

25 at 102-103°C is obtained.

 $[\propto]_D = -4.1^{\circ} \text{ (c=1\%, MeOH); } [\sim]_{436} = -17.4^{\circ} \text{ (c=1\% MeOH)}$ Example 3

R(-) 2-(3-benzoylphenyl)propionic acid S(-) 3-(4-phenyl-piperazin-1-yl)propane-1,2-diol salt

30 R(-) 2-(3-benzoylphenyl)propionic acid R(+) 3-(4-phenyl-

#### piperazin-1-yl)propane-1,2-diol salt

By salification of a solution of 1 g of R(-) 2-(3-benzoylphenyl)propionic acid in 10 ml acetone heated to 40°C with S(-) 3-(4-phenylpiperazin-1-yl)propane-1,2-diol and followed by cooling at room temperature a precipitate is separated which is filtered and dried in vacuo at 50°C (20 mm Hg) and gives R(-) 2-(3-benzoylphenyl)propionic S(-) 3-(4-phenylpiperazin-1-yl)propane-1,2-diol salt melting at 107-109°C.

- 10  $[\omega]_D = -20.4^\circ$  (c=1%, MeOH);  $[\omega]_{436} = -39.5^\circ$  (c=1% MeOH)

  Operating in a similar manner, by salifying with R(+)

  3-(4-phenylpiperazin-1-yl)propane-1.2-diol, the R(-) 2-(3-benzoylphenyl)propionic acid R(+) 3-(4-phenylpiperazin-1-yl)propane-1.2-diol salt melting at 118-120°C is obtained.
- 15  $\left[ \swarrow \right]_{D} = -1.5^{\circ} \text{ (c=1\%, MeOH); } \left[ \swarrow \right]_{436} = -3^{\circ} \text{ (c=1\% MeOH)}$ Example 4

## R(-) 2-(3-benzoylphenyl)propionic acid L-arginine salt S(+) 2-(3-benzoylphenyl)propionic acid L-arginine salt

A solution of 0.6 g L-arginine in 1 ml boiling water under gentle stirring is added to a solution of 1.02 g of R(-) 2-(3-benzoylphenyl)propionic acid in 10 ml acetone, heated to 40-45°C; a solid is separated which is filtered hot gives 1.3 g of R(-) 2-(3-benzoylphenyl)propionic acid L-arginine salt melting at 75°C.

The following filtration of the solid gives 1.12 g of S(+) 2-(3-benzoylphenyl)propionic acid L-arginine salt, melting at 85°C.

 $[\alpha]_D = +1.6^{\circ} \text{ (c=1\%, MeOH); } [\alpha]_{436} = -3.7^{\circ} \text{ (c=1\% MeOH)}$ 

5 Example 5

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S(+) 2-(3-benzoylphenyl)propionic acid L-lysine salt .1/4  $_{2}$ O Grams 0.28 of L-lysine dissolved at 80°C in 0.3 ml of distilled water are added to a solution of 0.5 g S(+) 2-(3-benzoylphenyl)propionic acid (o.p. > 90%;  $_{1}$  = + 50° in dichloromethane) in 10 ml isopropil alcohol, heated at 40°C.

The so obtained solution is left under stirring; for cooling, an oil is separated which, while it solidifies, is dispersed under stirring, forming a fine crystalline powder. The precipitate is filtered, first washed with isopropyl alcohol and then with ethyl alcohol.

Grams 0.55 g of L-lysine salt of S(+) 2-(3-benzoylphenyl) propionic acid .1/4  $H_2^0$  (o.p.of the acid > 99%) is obtained, melting at 147-149°C, the X-ray diffraction spectrum of which is shown in Fig.2.

20  $(H_2O)K.F.: 1\% + 0.3\%$   $[\alpha]_D = -0.3^{\circ} (c=1\%, MeOH); [\alpha]_{436} = -9.1^{\circ} (c=1\%, MeOH)$ Example 6

S(+) 2-(3-benzoylphenyl)propionic acid D-lysine salt .H<sub>2</sub>O

Grams 0.32 of D-lysine monohydrate dissolved at 80°C in 0.3 ml

of distilled water are added under stirring to a solution of 0.5 g of S(+) 2-(3-benzoylphenyl)propionic acid (o.p. > 90%;

[G]<sub>D</sub> = + 50° in dichloromethane) in 5 ml absolute ethyl alcohol.

It is diluted with 5 ml of absolute ethyl alcohol under continuous stirring and kept at 0°C for 5 hours. The precipitate which is formed is filtered and washed with

absolute ethyl alcohol. After drying 0.5 g of D-lysine salt monohydrate of S(+) 2-(3-benz ylphenyl)propionic acid (o.p > 99%) is obtained, melting at 108-110°C, the x-ray diffraction spectrum of which is shown in Fig.3.

- (H<sub>2</sub>O)K.F.: 4% + 0.5%  $[x]_{D} = -10.1^{\circ} (c=1\%, MeOH); [A]_{436} = -29.1^{\circ} (c=1\% MeOH)$ Example 7
  - S(+) 2-(3-benzoylphenyl)propionic acid phenylpiperazin-1-yl)propane-1,2-diol salt
- 10 Grams 0.5 of (+) 3-(4-phenylpiperazin-1-yl)propane-1,2-diol are added under stirring to a solution of 0.55 g of S(+)2-(3-benzoylphenyl)propionic acid (o.p.  $\gg 90\%$ ;  $[\alpha]_{D} = + 50^{\circ}$  in dichloromethane) in 5 ml of actione, heated at about 40°C. It is left to cool at room temperature to facilitate a slow
- separation of the salt. After 3 hours a crystalline 15 precipitate consisting of3-(4-phenylpiperazin-1-yl)propane-1,2-diol salt of the S(+) 2-(3-benzoylphenyl) propionic acid (o.p. > 99%) and melting at 107-109°C, is separated by filtration.
- $[\alpha]_{11} = +20^{\circ}, 4; [\alpha]_{436} = +38^{\circ}, 0 (c=1\% \text{ MeOH})$ 20 Example 8

consisting

- S(+) 2-(3-benzoylphenyl)propionic acid phenylpiperazin-1-y1)propane-1,2-diol salt
- Grams 5 of (-) 3-(4-phenylpiperazin-1-yl)propane-1,2-diol are added under stirring to a ::olution of 0.55 g of S(+)25 2-(3-benzoylphenyl)propionic acid (o.p.  $\gg 90\%$ ;  $\left[\alpha\right]_{D} = + 50^{\circ}$  in dichloromethane) in 5 ml of acrtone, heated at about 40°C. It is left to cool at room temperature to facilitate a slow separation of the salt. After 3 hours a crystalline 30 precipitate

0.67

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3-(4-phenylpiperazin-1-yl)propane-1,2-diol salt of the S(+) 2-(3-benzoylphenyl) propionic acid (o.p. > 99%) and melting at 118-120°C, is separated by filtration.

 $[\alpha]_D = +1.2 \text{ (c=1% MeOH)}; [\alpha]_{436} = +2.3 \text{ (c=1% MeOH)}$ 

The crystallographic analysis of the tested compounds has been carried out using a PWl 700 Automated Power Diffratometer System apparatus.

#### Example 9

By re-crystallization from acetone of each of the
enantiomerically rich salts obtained according to preparations
1 and 2 the following diastereoisomerically pure salts are
obtained:

- (2S,2'S) 3'-(4'phenylpiperazin-1'-yl)propane-1',2'-diol 2-(3-benzoylphenyl)propionate, melting at 118-120°C [~]<sub>D</sub> = +1.2° (MeOH);
- (2R,2'R) 3'-(4'phenylpiperazin-1'-yl)propane-1',2'-diol 2-(3-benzoylphenyl)propionate, melting at 118-120°C  $[\propto]_D$  = +1.5° (MeOH);
- (2R,2'S) 3'-(4'phenylpiperazin-1'-yl)propane-1',2'-diol 2-(3-benzoylphenyl)propionate, melting at 107-109°C [ $\omega$ ]<sub>D</sub> = +20.4° (MeOH);
  - (2S,2'R) 3'-(4'phenylpiperazin-1'-yl)propane-1',2'-diol 2-(3-benzoylphenyl)propionate, melting at 107-109°C  $[\omega]_D$  = +20.4° (MeOH).

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#### CLAIMS

1. A salt of an enantiomer selected from S(+) and R(-) 2-(3-benzoylphenyl)propionic acid with an organic base selected from the group consisting of tris-(hydroxymethyl)aminomethane, L-lysine, D-lysine,

- 5 L-arginine, (S) 3-(4-phenylpiperazin-1-yl)propane- 1,2-diol and (R) 3-(4-phenylpiperazin-1-yl)propane-1,2-diol.
  - 2. S(+) 2-(3-benzoylphenyl)propionic acid L-lysine salt .1/4  $H_2^0$  having the diffraction characteristics which are listed as follows:

10	Peak	D space	I/Imax
	n°	(ang.)	(%)
	1	12.7720	12.21
	2	10.7535	7.98
	3	10.0828	4.33
	4	8.5129	14.38
	5	7.4181	92.70
15	6	7.0794	6.39
	7	6.6192	25.19
	8	6.3171	29.46
	9 .	6.1648	37.62
	10	5.9455	54.18
	11	5.7515	78.94
	12	5.7247	71.21
20	13	5.3841	50.94
	14	5.2264	15.82
	15	5.0498	59.79
	16	4.4683	100.00
	17	4.4133	98.52
	18	4.3313	32.35
	19	4.2638	32.35
	20	4.1649	27.87
25	21	4.1395	20.57
	22	3.9880	86.37
	23	3.8241	22.29
	24	3.7250	18.60
	25	3.6567	32.77
	26	3.6302	33.20
			55.20

30

	Peak	D space	I/Imax
	n°	(ang.)	(%)
	27	3.5272	39.47
	28	3.4374	19.90
	29	3.3052	19.90
	30	3.1667	25.19
_	31	3.1134	19.24
5	32	2.9534	11.95
	<b>3</b> 3	2.8460	4.65
	34	2.7126	6.58
	<b>35</b> .	2.6011	7.16
	36	2.4886	3.74
	37	2.3855	5.31
	38	2.3146	3.88
10	39	2.1364	2.32
	40	1.9261	1.42

3. S(+) 2-(3-benzoylphenyl)propionic acid D-lysine salt monohydrate having the diffraction characteristics given as follows:

15	Peak	D space	I/Imax	
	n°	(ang.)	(%)	
	1	9.7956	19.60	
	2	9.1056	15.88	
	3	8.2476	62.67	
	4	7.1854	11.46	
20	5	6.5400	38.74	
20	6	5.8402	9.77	
	7	5.3874	25.00	
	8	5.2549	20.77	
	9	4.9926	53.17	
	10	4.9096	100.00	
	11	4.7367	63.92	
	12	4.6074	61.04	
25	13	4.4440	32.53	
	14	4.4155	31.93	
	15	4.3266	64.33	
	16	4.1829	26.05	
	17	4.1172	29.91	
	18	4.0300	28.78	
	19	3.8120	48.71	

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Peak		D space	I/Imax	
n°		(ang.)	(%)	
20		3.6582	24.74	-
21		3.4670	11.11	
22		3.2828	22.71	
23		3.2208	11.29	
24		3.1389	12.18	
25		3.0527	13.29	
26		2.8978	8.06	
27		2.7561	9.60	
28		2.5991	8.66	
29		2.5130	4.56	
30		2.3760	4.67	
31		2.3255	0.69	
32		2.1000	3.92	
33		2.0117	1.76	
34		1.9626	1.56	
35		1.8935	1.70	
4.	S(+)	2-(3-benzoylphenyl	)propionic ac	- :id (
3-(4-	phenylpip	erazin-1-yl)propane-1	,2-diol salt.	
5.	S(+)	2-(3-benzoylphenyl	)propionic ac	id (
3-(4-1	phenylpip	erazin-l-yl)propane-l	,2-diol salt.	
6. R(-	-) 2-(3-ь	enzoylphenyl)propioni	c acid L-Lysine s	salt havi
	iffractio	n characteristics whi	ch are listed as	follows:
Peak		D space	I/Imax	
n°		(ang.)	(%)	
1		15.4982	3.86	-
2		10.1668	18.84	
3		9.3855	15.21	
4		8.4662	53.32	
5		7.3704	8.26	
6		6.7028	32.37	
7		0.0010	02.07	

6.0016

5.4910

5.3454

4.9982

4.8060

4.6883

5.92

20.40

18.33

100.00

57.24

58.14

	Peak n°	D space (ang.)	I/Imax (%)
	13	4.3906	60.85
	14	4.1883	33.38
	15	3.8515	43.92
5	16	3.7074	26.94
	17	3.4962	10.41
	18	3.3109	22.02
	19	3.1711	12.38
	20	3.0818	13.01
	21	2.9072	6.81
	22	2.7841	7.93
	23	2.6173	6.51
10	24	2.5279	4.84
	25	2.3990	5.78
	26	2.3419	3.41
	27	2.1063	3.10

- 7. R(-) 2-(3-benzoylphenyl)propionic acid D-lysine salt.
- 8. R(-) 2-(3-benzoylphenyl)propionic acid tris-hydroxymethylmethylamonium salt.
  - 9. S(+) 2-(3-benzoylphenyl)propionic acid tris-hydroxymethyl-methylamonium salt.
  - 10. R(-) 2-(3-benzoylphenyl)propionic acid S(-) 3-(4-phenylpiperazin-1-yl)propane-1,2-diol salt.
- 20 11. R(-) 2-(3-benzoylphenyl)propionic acid R(+) 3-(4-phenylpiperazin-1-yl)propane-1,2-diol salt.
  - 12. R(-) 2-(3-benzoylphenyl)propionic acid L-arginine salt.
  - 13. S(+) 2-(3-benzoylphenyl)propionic acid L-arginine salt.
- 14. A process for obtaining pure diastereoisomeric salts of
  (R) or (S) 2-(3-benzoylphenyl)propionic acid with R or S
  4-(3-phenylpiperazin-1-yl)propane-1,2-diol by fractional
  crystallization of the diastereoisomeric mixtures of salts.
  - 15. A process for the preparation of the salts of claim 1, characterized in that an enantiomeric form selected from R(-)
- 30 2-(3-benzoylphenyl)propionic acid and S(+)

2-(3-benzoylphenyl)propionic acid is salified in a suitable solvent with an organic achiral base such as tris-(hydroxymethyl)aminomethane or an organic chiral base selected from the group consisting of L-lysine, D-lysine, L-arginine, (R) 3-(4-phenylpiperazin-1-yl)propane-1,2-diol and

- (S) 3-(4-phenylpiperazin-1-yl)propane-1,2-diol.
  16. A pharmaceutical composition having anti-inflammatory activity, characterized by the fact that it contains as active
- nrinciple a therapeutically effective quantity of one or more compounds according to the claims 1-5 in admixture with suitable pharmaceutically acceptable eccipients.
  - 17. A pharmaceutical composition having analysis activity, characterized in that it contains as active principle one or more compounds according to claims 6-13 in admixture with suitable pharmaceutically acceptable excipients.
  - 18. A pharmaceutical composition according to claim 17, characterized in that the active principle is R(-) 2-(3-benzoylphenyl)propionic acid L-lysine salt.
- 19. A pharmaceutical composition according to claim 16, 20 characterized in that the active principle is S(+) 2-(3-benzoylphenyl)propionic acid L-lysine salt 1/4 H<sub>2</sub>O.

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FIGURE 1

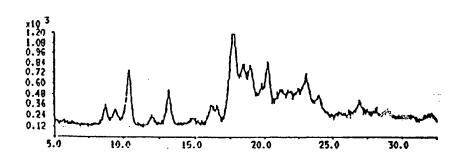
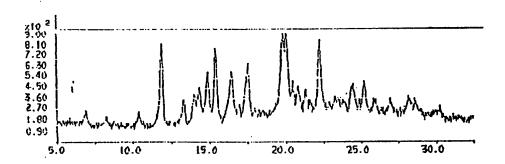


FIGURE 2

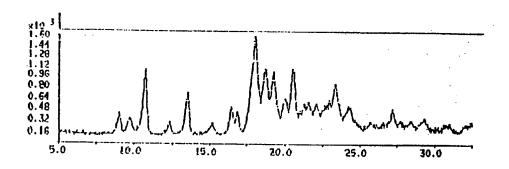


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FIGURE 3



## INTERNATIONAL SEARCH REPORT

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Minimum o IPC 5	documentation searched (classification system followed by classifi CO7C CO7D A61K CO7B	ication symbols)	
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Special cat  'A' docume conside  'E' earlier afiling of  'L' docume which citation  'O' docume other n  'P' docume later th	ent defining the general state of the art which is not ered to be of particular relevance document but published on or after the international date ent which may throw doubts on priority claim(s) or is cited to establish the publication date of another n or other special reason (as specified) ent referring to an oral disclosure, use, exhibition or	"T' later document published after the int or priority date and not in conflict we cited to understand the principle or t invention "X' document of particular relevance; the cannot be considered novel or canno involve an inventive step when the difference of the cannot be considered to involve an in document is combined with one or ments, such combination being obvicin the art.  "&" document member of the same pater.  Date of mailing of the international se	cernational filing date ith the application but heory underlying the claimed invention t be considered to occurrent is taken alone c claimed invention nventive step when the hore other such docu- ous to a person skilled t family
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Name and m	nailing address of the ISA  European Patent Office, P.B. 5818 Patentiaan 2  NL - 2280 HV Rijswijk  Tel. (+31-70) 340-2040, Tx. 31 651 epo ni, Pax (+31-70) 340-3016	Authorized officer  Klag, M	· · · · · · · · · · · · · · · · · · ·

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